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### Pre University Courses

Chemistry Notes for class 12 Chapter 2 Solutions Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution. Solute and Solvent

### Chemistry Notes for class 12 Chapter 2 Solutions

= 12.00 mmol - 5.40 mmol Unreacted ethanoic acid = 6.60 mmol Since 18.00 mL of NaOH(aq) was added to 20.00 mL of ethanoic acid solution, the total volume is now 38.00 mL. Solution: The concentration of unreacted ethanoic acid, [HC 2H 3O 2(aq)], can be determined: 23 2 23 2 HC H O (aq) 23 2 HC H O (aq) 23 2 [HC H O (aq)] (6.60 mmol) = 38.00 mL

### Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547

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Gary S. Thorpe has taught AP Chemistry, College Chemistry, and gifted programs for over 30 years. Recipient of numerous awards in teaching. He is currently on staff at Beverly Hills High School, Beverly Hills, California. Author's Acknowledgments I would like to thank my wife, Patti, and my two daughters,