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Electrochemical production of H₂O₂ from O₂ is a promising alternative to the energy-intensive anthraquinone process that is currently used as an industry standard. Although most research on the oxygen reduction reaction (ORR) has focused on the 4-electron pathway to water relevant to fuel cells, the 2-electron ORR to produce H₂O₂ is also of significant commercial interest. The first half of ...

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Table 4.1 presents the selected ORR processes and their corresponding thermodynamic electrode potentials at standard conditions. All the potentials are against the standard hydrogen potential, which is defined as zero in 1.0 M proton aqueous solution at any temperature and 1.0 atm hydrogen gas pressure. In this table, two cases of electrolyte solution are selected: one is the acidic aqueous ...

Electrochemical Oxygen Reduction Reaction - ScienceDirect

Zairan Liu, Yuanming Deng, Li Wang, Jiaoning Tang, A Facile Topochemical Preparation of Ni-Fe LDH Nanosheets Array on Nickel Foam Using In Situ Generated Ni²⁺ for Electrochemical Oxygen Evolution, Journal of The Electrochemical Society, 10.1149/1945-7111/ab72ee, 167, 4, (046502), (2020).

2D Layered Double Hydroxides for Oxygen Evolution Reaction ...

Electrochemistry, a branch of chemistry, went through several changes during its evolution from early principles related to magnets in the early 16th and 17th centuries, to complex theories involving conductivity, electric charge and mathematical methods. The term electrochemistry was used to describe electrical phenomena in the late 19th and 20th centuries.

History of electrochemistry - Wikipedia

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