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Chemistry: Form WS6.4.1A Name ____ PHASES OF MATTER Date ____ Period ____ Using the Combined Gas Law If a balloon is pulled over the neck of a flask, and the setup is placed on a hot plate, the balloon blows up as it heats up. This happens even though no additional air can get into the balloon.

Using the Combined Gas Law - Evan's Regents Chemistry Corner

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Ui'k ,f (Continued l') e y (fpe4) >j '// t / 3 Cf Form WS6, 3. 1A nLt,rf: Phase Changes PHASES OF MATTER 4. Page 2 Why do substances change phase as the temperature changes? Refer to the distance between particles and the force of f Le_ attraction between them in your answer.

WS6 - studylib.net

Chemistry: Form WS6.4.3A Name ____ PHASES OF MATTER Date ____ Period ____ Graham's Law A helium balloon bobs up and down, tethered by a string and buoyed by the air. The helium keeps it afloat. In a few days it will fall. The helium escapes from the balloon slowly through microscopic holes.

Graham's Law - Evan's Regents Chemistry Corner

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Chemistry: Form WS6.3.1A Name ____ PHASES OF MATTER Date ____ Period ____ Phase Changes Heating a substance in a given phase generally causes the temperature to increase. This causes the particles to move faster and collide harder. Harder, faster collisions cause particles to rebound harder moving them further apart. ...

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Chemistry: Form WS6.5.1A Name ____ PHASES OF MATTER Date ____ Period ____ The Ideal Gas How do gases behave? In order to study gases, chemists have devi sed a model. The model is called an ideal gas (a gas which explains the behavior of all gases).

The Ideal Gas - Evan's Regents Chemistry Corner

Chemistry: Form WS6.1.2A Name ____ PHASES OF MATTER Date ____ Period ____ Specific Heat The sun beat down on the beach, reflecting off the smooth surface of the lake and the bright white sand. A beach-goer stepped eagerly off the blanket anticipating the relief of the cool water. Yipes!!! The sand was hot!

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Phases of matter change by gaining or losing energy.. Heat decides the state of matter. Adding heat or taking off heat brings in change of state/ phase. Soild -----> Liquid This is called melting process (solid needs to be heated, example melting of butter, solid on taking up heat changes to liquid)

Phases of Matter - Chemistry | Socratic

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Example Summary Paragraph

f. Form N5 . 5A CHEMICAL FORMULAS AND EQUATIONS Conservation of Mass Answer the questions below by circling the number of the correct response 1. If 46 g of X combines with 16 g of Y to form Z, how much Z is formed? (1)30g (2)2.9g (3)724g (4)62g 2. The formula mass of sulfuric acid (H₂SO₄) is (1) 94amu, (2) 98 amu, (3) 50 amu, (4) 192 amu 3.

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