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Rate = $k [\text{CH}_3\text{OCH}_3]^{3/2}$ The rate of reaction is followed by increase in pressure in a closed vessel, so the rate can also, be expressed in terms of the partial pressure of dimethyl ether, i.e.,
Rate = $k (P_{\text{CH}_3\text{OCH}_3})^{3/2}$

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a) Ethane in staggered conformation has C₃-rotational axis passing through the C-C bond. It has three C₂-rotational axes perpendicular to the principal C₃-rotational axis and through the midpoint of the C-C bond. It has not any h-plane. But it has 3 d-planes containing

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two carbons and anti two hydrogens. Hence the point group of ethane in the staggered conformation will be D_{3d} .

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